

Original Research

# Effects of $\text{Cu}^{2+}$ and $\text{Hg}^{2+}$ on Growth and Photosynthesis of Two *Scenedesmus* Species

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## Abstract

$\text{Cu}^{2+}$  and  $\text{Hg}^{2+}$  are two typical contaminants. Previous studies on toxicity of  $\text{Cu}^{2+}$  and  $\text{Hg}^{2+}$  to green algae mainly employed concentrations higher than environmental levels. Since the results varied among different strains of the same species, toxicity assessment using local green alga strains might be more accurate for revealing risks of  $\text{Hg}^{2+}$  and  $\text{Cu}^{2+}$  to local environments. In the present study, *Scenedesmus quadricauda* and *Scenedesmus acutus* were isolated from the Xin'an River in Huangshan City, China. Both were treated with 0.01-0.15 mg/L  $\text{Hg}^{2+}$  or 0.5-10 mg/L  $\text{Cu}^{2+}$ . The results showed that  $\text{Hg}^{2+}$  and  $\text{Cu}^{2+}$  were highly toxic. Treatment with 0.1 mg/L  $\text{Hg}^{2+}$  completely inhibited growth of *S. acutus* and 0.15 mg/L  $\text{Hg}^{2+}$  inhibited growth of *S. acutus*, but no significant changes were observed in contents of photosynthetic pigments and chlorophyll fluorescence parameters, suggesting that toxicity of  $\text{Hg}^{2+}$  might not be due to inhibition on photosynthesis. Treatments with 0.5 mg/L  $\text{Cu}^{2+}$  depressed cell growth, and higher levels of  $\text{Cu}^{2+}$  decreased contents of photosynthetic pigments (chl-a, car or chl-b) in *S. quadricauda* and *S. acutus*. Moreover, *S. quadricauda* might be more sensitive to heavy metal treatments than *S. acutus*. These results should be useful for evaluating environmental risks of  $\text{Hg}^{2+}$  and  $\text{Cu}^{2+}$  to Huangshan City.

**Keywords:** alga density, chlorophyll, fluorescence, *Scenedesmus quadricauda*, *Scenedesmus acutus*

## Introduction

$\text{Hg}^{2+}$  and  $\text{Cu}^{2+}$  are two typical contaminants. Environmental pollution of  $\text{Hg}^{2+}$  and  $\text{Cu}^{2+}$  have been observed globally.  $\text{Hg}^{2+}$  is high toxicity to human beings and environments due to its bioaccumulation and biomagnification within trophic levels [1].  $\text{Hg}^{2+}$  could

be transformed to methylmercury, which is even more toxic [2].  $\text{Hg}^{2+}$  and derivatives can cause severe disorders to organisms, including neurological, immunological, cardiac, motor, reproductive and genetic toxicity, and also is associated with Alzheimer's, Parkinson's, autism, lupus, and amyotrophic lateral sclerosis [3,4]. Recent molecular studies have revealed that  $\text{Hg}^{2+}$  treatment decreased many critical processes, including ferric iron binding, antioxidant activity, cellular homeostasis, and glutathione metabolism in copepod

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[5]. Metabolomics analyses revealed that  $\text{Hg}^{2+}$  negatively affected ion-osmoregulatory, damaged cell membrane and induced hypoxic stress in fish [1].

$\text{Cu}^{2+}$  is an essential micronutrient playing an important role in many metabolic processes, as a cofactor for enzymes. Excess  $\text{Cu}^{2+}$  initiates oxidative damage and interferes with important cellular components in organisms, which further leads to abnormal  $\text{Cu}^{2+}$  metabolism and neurodegenerative changes [6].

Algae are important components in aquatic ecosystems. Influences and bioaccumulation of heavy metals in alga cells will affect high trophic levels and the whole ecosystem [7]. Toxicity assessments of  $\text{Hg}^{2+}$  and  $\text{Cu}^{2+}$  to green algae demonstrated that  $\text{Hg}^{2+}$  is the most toxic heavy metal, followed by  $\text{Cu}^{2+}$  [8]. Both of them inhibited growth, cell permeability, photosynthesis and/or nitrogen fixation in algae [9-14], and altered alga communities [15]. These effects were influenced by environmental conditions and water chemistry, i.e., temperature [16], water hardness and alkalinity [17]. However, most studies tested the toxicity of  $\text{Hg}^{2+}$  and  $\text{Cu}^{2+}$  at concentrations ranging from 0.5 mg/L to 100 mg/L. Obviously, these concentrations were much higher than those in real environments, which could not comprehensively reveal environmental risks of  $\text{Hg}^{2+}$  and  $\text{Cu}^{2+}$  pollution in nature.

Toxicity of pollutants to green algae varied among species [14] and strains [17, 18]. Investigations on more species and strains from local water bodies might more accurately evaluate pollution on local environments [17]. *S. quadricauda* and *S. acutus* are two dominant alga species in aquatic environments. In the present study, *S. quadricauda* and *S. acutus* isolated from the Xin'an River (Huangshan, China) were employed as model organisms. *S. quadricauda* and *S. acutus* were treated with trace amounts of  $\text{Hg}^{2+}$  and  $\text{Cu}^{2+}$ . Thereafter, changes of growth indices, chlorophyll contents and photosynthetic parameters were compared. These results would be useful for evaluating risks of trace  $\text{Hg}^{2+}$  and  $\text{Cu}^{2+}$  to local environments.

## Materials and Methods

*S. quadricauda* and *S. acutus* were isolated from the Xin'an River in Huangshan City, P. R. China, and then cultured in 500 ml flasks containing 300 ml of BG-11 medium [19] at  $25 \pm 1^\circ\text{C}$ . The photoperiod was 12 h: 12 h (light: dark) with light intensity of 6,000 lux. Algae were manually shaken three times per day.

For treatments with heavy metals, algae at the exponential growth stage were harvested and pooled. Alga density was determined using a hemocytometer and then the initial alga density was adjusted to  $1 \times 10^5$  cells/mL. Next,  $\text{HgCl}_2$  and  $\text{CuCl}_2$  (analytic grade) were used as the metal sources. Four concentrations of  $\text{Hg}^{2+}$  (0.01, 0.05, 0.1, 0.15 mg/L) and  $\text{Cu}^{2+}$  (0.5, 1, 5, 10 mg/L) were prepared. BG-11 media without additional  $\text{Cu}^{2+}$  or  $\text{Hg}^{2+}$  were included as the control. Each assay

was repeated three times independently. Alga density was monitored every 24 hours for 10 days to calculate population growth rate.

Contents of photosynthetic pigments were determined on day 10. Briefly, 150 ml of algae solution were sampled from each treatment and then vacuum filtrated on 0.22- $\mu\text{m}$  filter membrane (Whatman GF/F). Algae were grounded in 10 ml of 95% ethanol and then extracted at  $4^\circ\text{C}$  in dark for 12 hours. After centrifuging at 5,000 rpm for 10 min, absorbance of supernatants at 665 nm, 649 nm and 647 nm was determined using a spectrophotometer. Contents of chlorophyll a (chl-a), chlorophyll b (chl-b) and carotenoids (car) were calculated according to Yang et al. [20].

On day 5, chlorophyll fluorescence parameters were determined. Briefly, 100 ml of algae solution was collected and then placed in the dark for 15 min. Afterward, chlorophyll fluorescence parameters, including maximal photochemical efficiency of PSII

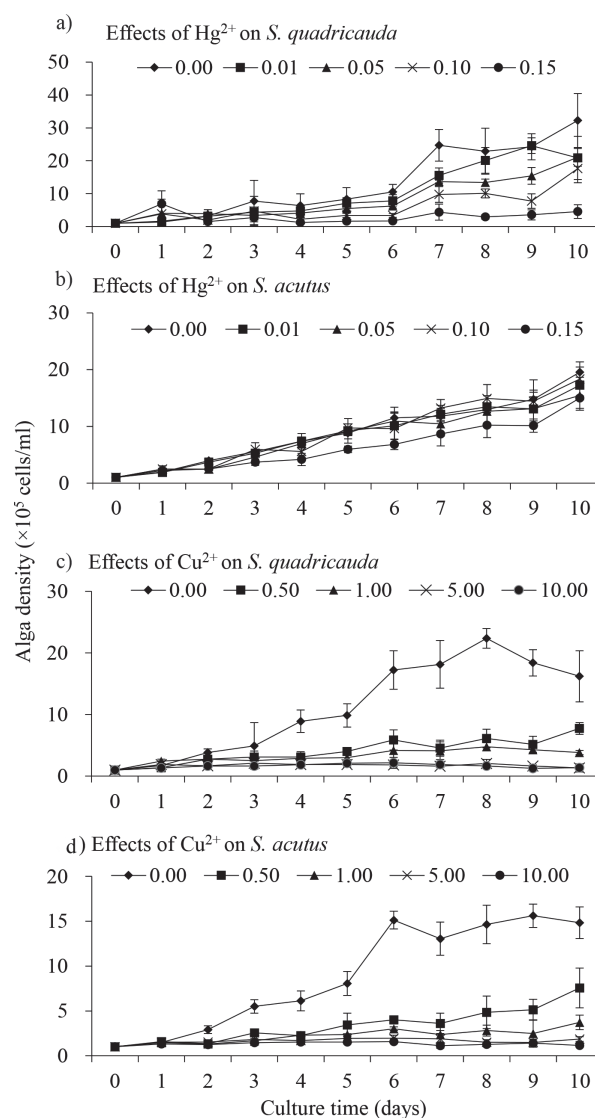


Fig. S1. Growth curves in treatments with  $\text{Hg}^{2+}$  (mg/L) and  $\text{Cu}^{2+}$  (mg/L). Data represent mean  $\pm$  SD.

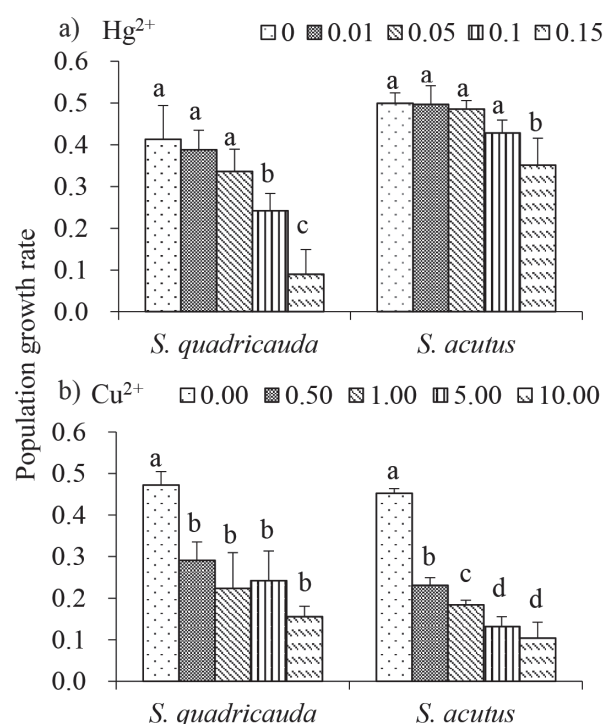


Fig. 2. Effects of  $\text{Hg}^{2+}$  (mg/L) and  $\text{Cu}^{2+}$  (mg/L) on population growth rate of *S. quadricauda* and *S. acutus* (mean $\pm$ SD). Different letters above bars represent significant differences.

( $F/F_m$ ), actual photochemical efficiency of PSII (yield), maximal relative electron transport rate ( $rETR_{max}$ ), initial slope rate ( $\alpha$ ) and half-saturation light intensity ( $I_k$ ) were determined using a phytoplankton fluorescence instrument (phyto-PAM, Walz-Germany).

After testing homogeneity of variance, the effects of  $\text{Cu}^{2+}$  or  $\text{Hg}^{2+}$  on each parameter were analyzed using one-way analysis of variance (ANOVA) followed by least significant difference (LSD) using SPSS 19.0.

## Results

### Effects of $\text{Cu}^{2+}$ and $\text{Hg}^{2+}$ on Algae Growth

Along with time, alga density increased gradually (Fig. S1), but population growth rate decreased with the elevation of  $\text{Hg}^{2+}$  and  $\text{Cu}^{2+}$  concentrations. Compared with the control, treatments with 0.1-0.15 mg/L  $\text{Hg}^{2+}$  and 0.5-10 mg/L  $\text{Cu}^{2+}$  significantly decreased population growth rate of *S. quadricauda*. Treatments with 0.15 mg/L  $\text{Hg}^{2+}$  and 0.5-10 mg/L  $\text{Cu}^{2+}$  significantly reduced population growth rate of *S. acutus* (Fig. 2).

### Effects of $\text{Cu}^{2+}$ and $\text{Hg}^{2+}$ on Contents of Photosynthetic Pigments

At all tested concentrations,  $\text{Hg}^{2+}$  did not significantly affect contents of chl-a, car and chl-b in either *S. quadricauda* or *S. acutus* ( $P > 0.05$ )

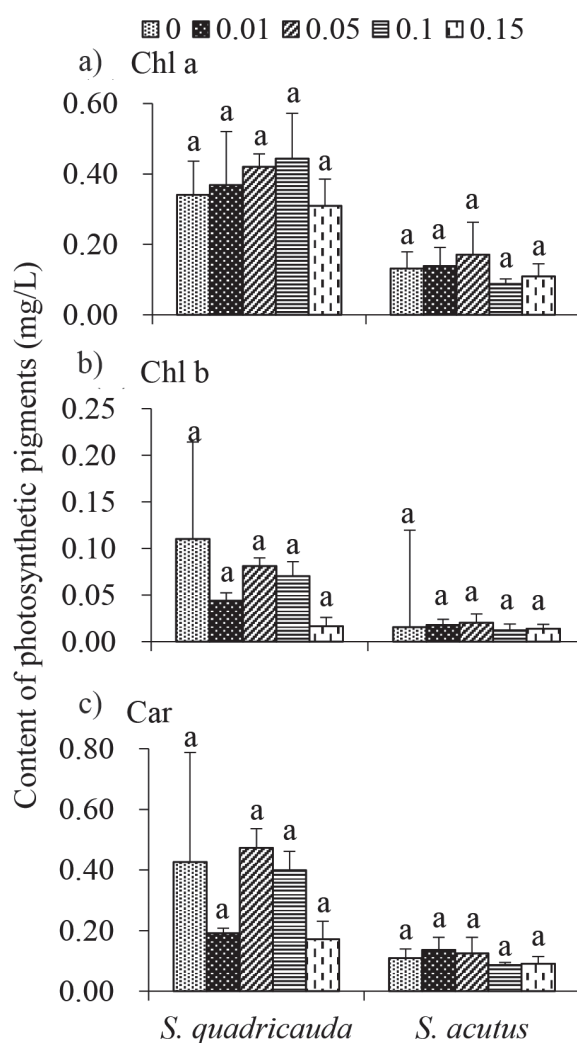


Fig. 3. Effects of  $\text{Hg}^{2+}$  (mg/L) on contents of photosynthetic pigments in *S. quadricauda* and *S. acutus* (mean $\pm$ SD). Different letters above bars represent significant differences.

(Fig. 3). In comparison, all these parameters were significantly affected by exposure to  $\text{Cu}^{2+}$ . Treatments with 0.5-10 mg/L  $\text{Cu}^{2+}$  significantly decreased contents of chl-a, chl-b and car in *S. quadricauda*, except car in treatment with 0.5 mg/L, which was not significantly different from the control. Treatment with 0.5 mg/L  $\text{Cu}^{2+}$  significantly increased contents of all pigments in *S. acutus*, but treatments with 5 and 10 mg/L  $\text{Cu}^{2+}$  significantly suppressed contents of chl-a and car (Fig. 4).

### Effects of $\text{Cu}^{2+}$ and $\text{Hg}^{2+}$ on Parameters of Chlorophyll Fluorescence in Algae

In response to treatments with 0.01-0.15 mg/L  $\text{Hg}^{2+}$ , no significant differences were detected in all tested chlorophyll fluorescence parameters in both *S. quadricauda* and *S. acutus* (Fig. 5).

In *S. quadricauda*, treatment with 0.5 mg/L  $\text{Cu}^{2+}$  significantly induced  $F_v/F_m$ , yield,  $rETR_{max}$  and  $I_k$ ,

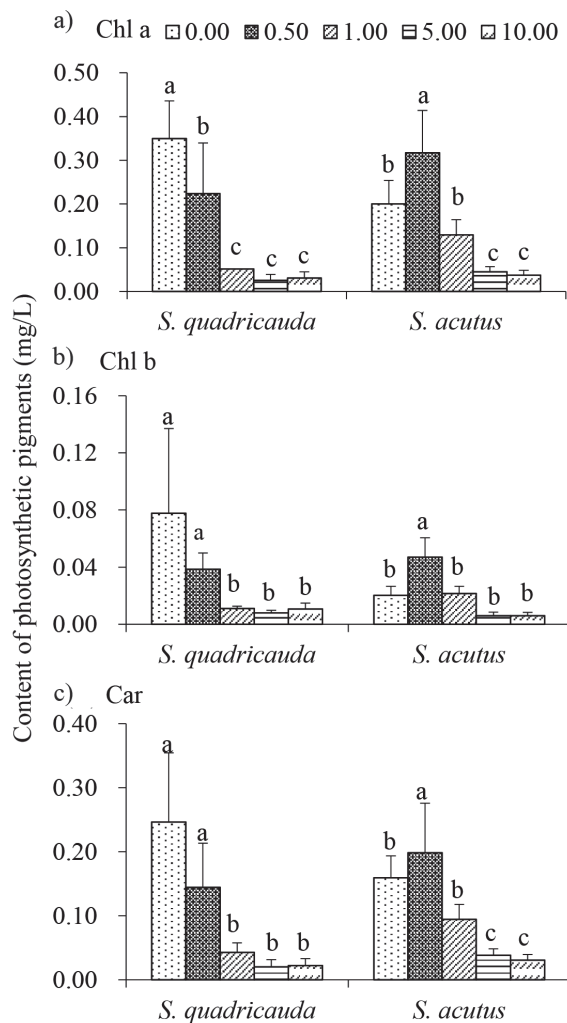


Fig. 4. Effects of  $\text{Cu}^{2+}$  (mg/L) on contents of photosynthetic pigments in *S. quadricauda* and *S. acutus* (mean $\pm$ SD). Different letters above bars represent significant differences.

which were then gradually declined with elevated concentration (Fig. 6).  $\alpha$  was not affected in treatments with 0.5 and 1 mg/L  $\text{Cu}^{2+}$ , but also declined at 1-10 mg/L  $\text{Cu}^{2+}$ . In contrast, effects of  $\text{Cu}^{2+}$  on chlorophyll fluorescence parameters of *S. acutus* were more complicated. Treatments with 0.5 and 1.0 mg/L  $\text{Cu}^{2+}$  significantly elevated  $\alpha$ ,  $F_v/F_m$  and yield. All these indices were significantly lower in treatment with 5 and 10 mg/L than those in the control. Treatment with 0.5 mg/L  $\text{Cu}^{2+}$  increased  $rETR_{\max}$  but did not affect  $I_k$ . Both indices were significantly reduced at 1-10 mg/L  $\text{Cu}^{2+}$  (Fig. 6).

## Discussion

$\text{Hg}^{2+}$  is highly toxic to green algae. Treatment with 0.1 mg/L  $\text{Hg}^{2+}$  completely inhibited growth and significantly decreased photosynthesis in *S. acutus* [21]. Similarly, in the present study, treatment with 0.1 mg/L  $\text{Hg}^{2+}$  significantly inhibited growth of *S.*

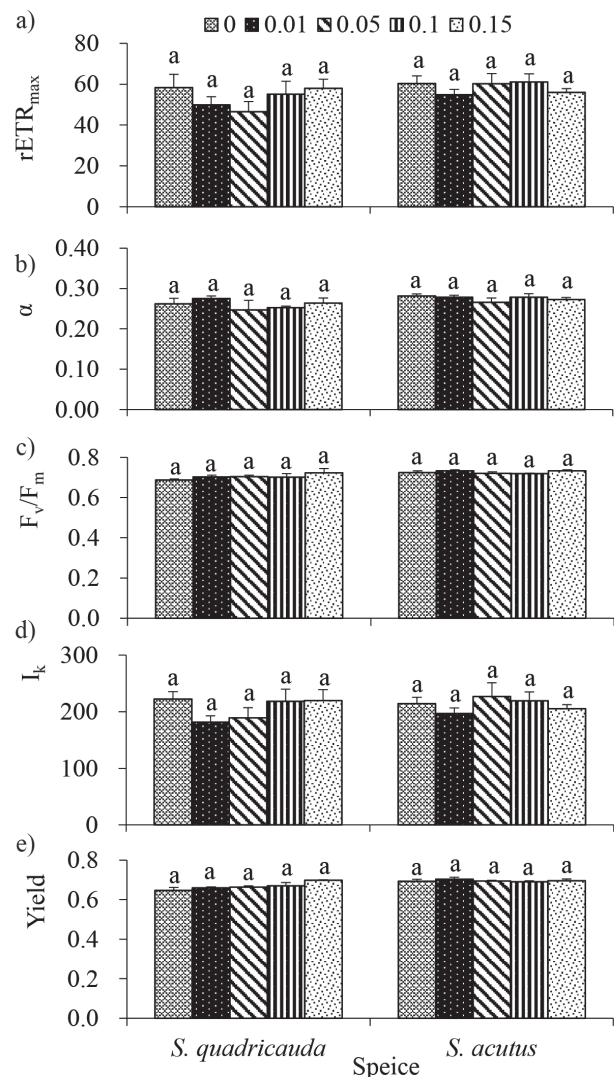


Fig. 5. Effects of  $\text{Hg}^{2+}$  (mg/L) on chlorophyll fluorescence parameters in *S. quadricauda* and *S. acutus* (mean $\pm$ SD). Different letters above bars represent significant differences.

*quadricauda* and 0.15 mg/L  $\text{Hg}^{2+}$  significantly inhibited growth of *S. acutus*, displaying toxicity of  $\text{Hg}^{2+}$  to environments.

To explore mechanisms underlying inhibition of  $\text{Hg}^{2+}$  to alga density, contents of photosynthetic pigments and changes of chlorophyll fluorescence parameters were monitored. No significant changes in content of photosynthetic pigments were detected in  $\text{Hg}^{2+}$  treatments, which seemed conflicted with the suppressed alga density. It has been revealed that treatment with  $\text{Hg}^{2+}$  increased cell size of live *T. weissflogii* cells [22]. Enlarged alga cells could synthesize more photosynthetic pigments, which might supplement loss of pigment content caused by reduced cell density.

As previously reported,  $\text{Hg}^{2+}$  decreased PSII quantum yield in *M. pteropus* [23], inhibited the transfer of excitation energy within phycobilisomes in *Spirulina platensis* [24] and decreased the capacity of photosystem to dissipate the excitation light energy via the photochemical pathway in *M. aeruginosa* [25].

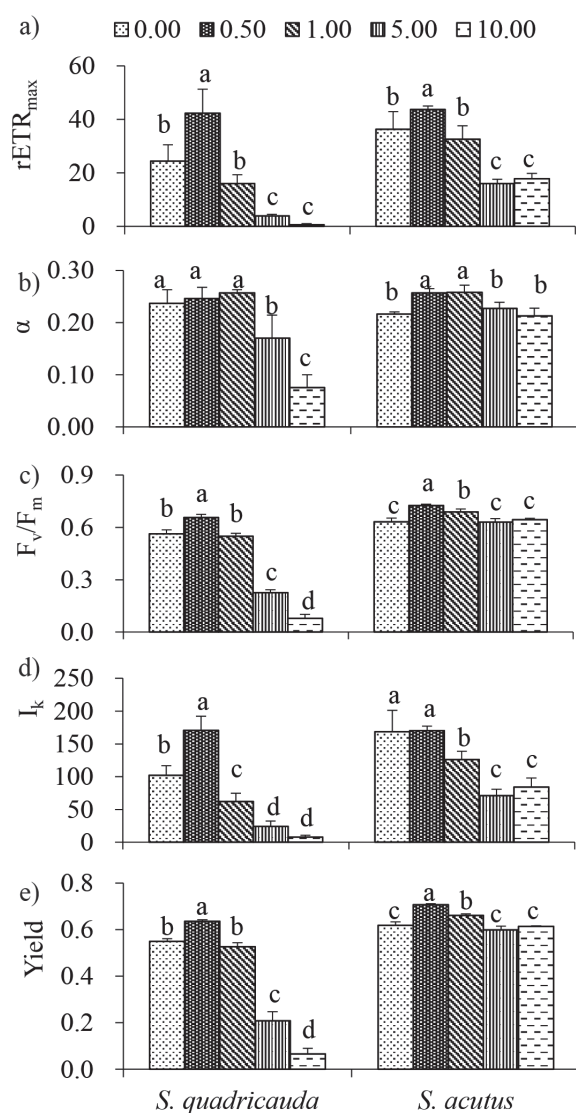


Fig. 6. Effects of  $\text{Cu}^{2+}$  (mg/L) on chlorophyll fluorescence parameters in *S. quadricauda* and *S. acutus* (mean $\pm$ SD). Different letters above bars represent significant differences.

However, treatments with up to 370 nM  $\text{Hg}^{2+}$  almost did not affect chlorophyll fluorescence parameters in *C. vulgaris* and *P. biwae* [25]. In the present study, chlorophyll fluorescence parameters were not significantly affected by  $\text{Hg}^{2+}$  treatments, suggesting that the photosynthetic system was quite tolerant to  $\text{Hg}^{2+}$  in these two species.

Together with the unchanged contents of photosynthetic pigments, the present study demonstrated that inhibition of photosynthesis might not be the primary reason underlying toxicity of  $\text{Hg}^{2+}$  to growth of *S. acutus* and *S. quadricauda*. Several possibilities might explain the depressed alga density in response to  $\text{Hg}^{2+}$ , including inhibiting active transport of nutrients, nitrogen starvation [21], generation of reactive oxygen species (ROS) and oxidative damage [26]. However, more investigations are required to clarify this issue.

$\text{Cu}^{2+}$  is an essential element to the photosynthetic process. In the present, contents of chl-a, chl-b and car,  $rETR_{max}$ ,  $\alpha$ ,  $F_v/F_m$  and yield increased in *S. acutus* treated with 1 mg/L  $\text{Cu}^{2+}$ , suggesting a promotive effect of a low level of  $\text{Cu}^{2+}$  on photosynthesis. Similar effects were also reported in plant species, such as cereal crops [27,28]. Increased contents of pigments might be attributed to the promotion of  $\text{Cu}^{2+}$  on terpenoid biosynthesis [29].

Similar to previous reports [8,30], treatments with  $\text{Cu}^{2+}$  depressed cell growth and decreased contents of photosynthetic pigments (chl-a, car or chl-b) in *S. quadricauda* and *S. acutus*, demonstrating that high levels of  $\text{Cu}^{2+}$  triggered severe environmental concerns. A high level of  $\text{Cu}^{2+}$  blocked the electron transport and subsequently inhibited PSII activity [31], which further decreased  $F_v/F_m$  and adversely affected photosynthesis of algae [31-35]. In the present study, treatments with high levels of  $\text{Cu}^{2+}$  decreased contents of photosynthetic pigments,  $rETR_{max}$  and  $I_k$  in both *S. quadricauda* and *S. acutus*, suggesting that the photosynthetic process was inhibited. Inhibition of  $\text{Cu}^{2+}$  to pigment accumulation and retarded chlorophyll integration into the photosystems [36] through competing with  $\text{Mg}^{2+}$  [37] might explain these phenomena. Overall, these results together suggested that  $\text{Cu}^{2+}$  affected alga growth probably through a mediating photosynthesis process.

Sensitivity to heavy metals differed among alga species. In the present study, contents of photosynthetic pigments decreased or did not change in *S. quadricauda*, but increased in *S. acutus*, in response to treatment with 0.5 mg/L  $\text{Cu}^{2+}$ . When exposed to 0.1 mg/L  $\text{Hg}^{2+}$ , population growth rate decreased in *S. quadricauda*, but did not change in *S. acutus*. These results suggested that *S. quadricauda* might be more sensitive to heavy metal pollution than *S. acutus*.

## Conclusions

Both  $\text{Cu}^{2+}$  and  $\text{Hg}^{2+}$  significantly inhibited growth of *S. quadricauda* and *S. acutus*. Treatments with  $\text{Hg}^{2+}$  did not affect but treatments with  $\text{Cu}^{2+}$  significantly reduced contents of photosynthetic pigments and chlorophyll fluorescence parameters. Inhibition of photosynthesis might be a major reason underlying toxicity of  $\text{Cu}^{2+}$  to growth of *S. acutus* and *S. quadricauda*.

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## Conflict of Interest

The authors declare no conflict of interest.

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